Claims

1. (currently amended) A method [[of]] <u>for</u> producing heat energy, comprising the steps of:

providing a container for receiving an electrolyte composition, a cathode and an anode; forming an electrolyte composition comprising D₂O and sulfuric acid;

placing a sufficient amount of the electrolyte composition in the container to at least partially cover a <u>platinum or titanium metal</u> cathode made from a metal selected from the group eonsisting of palladium, platinum and titanium and to at least partially cover an inert anode situated inside the container;

connecting the cathode and anode to a source of electricity; and applying a current density across the cathode and anode of at least 0.55A/cm².

- 2. (canceled)
- 3. (currently amended) The method of claim 1 wherein, the electrolyte during the application of voltage, the electrolyte is held within a container and wherein said the container bounds a space above the electrolyte, [[said]] the space providing a region for [[the]] recombining [[of]] gases produced during the electrolysis.
- 4. (currently amended) The method of claim 1 wherein a catalyst is provided within [[said]] the region catalyzing the recombining of gases produced by the electrolysis.
 - 5. (canceled).
- 6. (currently amended) The method according to claim [[5]] $\underline{1}$ wherein the size of the cathode is about 1 cm².
- 7. (original) The method according to claim 1 wherein the cathode is made from titanium.

Page 2 of 5

- 8. (original) The method according to claim 1 wherein the inert anode is a platinum anode.
- 9. (original) The method according to claim 1 wherein the electrolyte composition consists essentially of D₂O and about 15% sulfuric acid by volume.
- 10. (currently amended) The method according to claim 9 wherein the cathode is made from palladium or a titanium metal cathode.
- 11. (currently amended) A method [[of]] <u>for</u> producing heat energy, comprising-the steps of:

providing a container for receiving an electrolyte composition, a cathode and an anode; forming an electrolyte composition comprising D₂O and sulfuric acid;

placing a sufficient amount of the electrolyte composition in a container to at least partially cover a <u>titanium metal</u> cathode made from a metal selected from the group consisting of nonhydride forming metals and to at least partially cover an inert anode situated inside the container;

connecting [[said]] the cathode and anode to a source of electricity; and applying a voltage of about 3.5 volts across the cathode and anode.

12. (currently amended) A method [[of]] <u>for</u> producing heat energy, comprising-the steps of:

providing a container for receiving an electrolyte composition, a cathode and an anode; forming an electrolyte composition consisting essentially of D₂O and 15% by volume sulfuric acid;

placing a sufficient amount of the electrolyte composition in a container to at least partially cover a palladium or titanium cathode and an inert anode situated inside the container, wherein the container bounds a space above [[said] the electrolyte composition;

connecting [[said]] the cathode and anode to a source of electricity; applying a voltage across the cathode and anode; and

providing a catalyst within the space above the electrolyte composition to catalyze the recombination of gases produced by the electrolyte.

Page 4 of 5